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Enantioselective Synthesis of Dihydropyridinones via NHC-Catalyzed Aza-Claisen Reaction

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ABSTRACT

N-Heterocyclic carbene catalyzed aza-Claisen annulations of enals or their α' -hydroxyenone surrogates with vinylogous amides afford dihydropyridinones. The reaction proceeds with a broad range of substrates, and no nitrogen protecting group is required.

NHC-catalyzed annulations of α , β -unsaturated aldehydes and various imine derivatives provide facile entry to a variety of N-heterocycles, often with exceptional enantioselectivity (Scheme 1). The rapid generation of structural and stereochemical diversity from a single class of starting materials and essentially one chiral N-heterocyclic carbene precursor, N-mesityl substituted chiral triazolium salt 1, make these reactions attractive approaches to scaffolds for further elaboration into bioactive compounds. The synthetic utility of the immediate annulation products, however, is diminished by the use of difficult to remove protecting groups. Furthermore, while some of the N-heterocycle forming annulations proceed with aliphatic and heteroaromatic substituents, the majority require aryl derivatives.

In this report we document NHC-catalyzed annulations of enals or their α' -hydroxyenone surrogates and readily prepared primary vinylogous amides. No nitrogen protecting group is employed, and the annulation delivers synthetically and medicinally important dihydropyridinone products suitable for direct biological assay or further elaboration into compounds of contemporary interest. The use of chiral N-heterocyclic carbene precursor 1 delivers the annulation products with good yields and enantioselectivies. The underlying mechanism of this cascade reaction features a novel NHC-catalyzed aza-Claisen rearrangement that evolved from our recent studies of NHC-catalyzed Coates—Claisen rearrangements.

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Scheme 1. NHC-Catalyzed Syntheses of N-Heterocycles

In 2010 we reported the enantioselective, NHC catalyzed synthesis of dihydropyranones⁶ from ynals and stable enols via catalytically generated α,β -unsaturated acyl azolium intermediates. A conceptually related reaction starting from α,β -unsaturated acyl fluorides and silyl enol ethers to deliver racemic dihydropyranones was reported just prior to our study by Lupton;⁷ related annulations were reported by Studer, 8 Xiao, 9 and You¹⁰ shortly after our work. As part of mechanistic studies establishing that the key C-C bond forming event occurs via an NHCcatalyzed variant of the Coates—Claisen rearrangement, 11 we found that the formation of a metastable hemiacetal from the $\alpha.\beta$ -unsaturated acyl azolium and the enol was key to the success of this reaction. Simultaneously, acyl azoliums have been established to be surprisingly reluctant to effect amidation reactions.12 Taken together, these observations suggested that stable, unsubstituted enamines would be suitable substrates for an aza-Claisen rearrangement without competing amide-bond formation.

The requisite enamines are either commercially available or prepared from 1,3-ketoesters.

Table 1. Optimization of Enantioselective Annulation of Enal^a

base or additive	ee %	base or additive	ee %
no base (3 days)	66	$i ext{-}\mathrm{Pr}_2\mathrm{NEt}$	80
K_2CO_3	68	$i ext{-} ext{Pr}_2 ext{NEt} + ext{NaBF}_4$	80
$t ext{-BuOK}$	73	$i ext{-} ext{Pr}_2 ext{NEt} + 4~ ext{Å MS}$	80
NMM	79	$i\text{-}\mathrm{Pr}_2\mathrm{NEt}$ at 23 °C	86

^a Reaction conditions: 0.1 M PhCH₃, 12 h at 40 °C unless noted.

Although the key α,β -unsaturated acyl azolium can be generated via an internal redox reaction of ynals and an NHC, ¹³ we choose to catalytically generate this species via an oxidation of the Breslow intermediate formed from the combination of enals and an NHC. Our studies began by examining the NHC-catalyzed annulation of enal 3 and enamine 4 in the presence of N-mesityl catalyst 1 (Table 1). Oxidant 2, reported by Kharasch¹⁴ and utilized by Studer, 8,12g was found to be the most effective for this reaction. The choice of base had a subtle effect. Amine bases such as DBU, NMM, and i-Pr2NEt were found to give higher enantioselectivity than inorganic bases (K₂CO₃, t-BuOK, or Cl⁻). Additives such as molecular sieves 9b or NaBF₄ 10 did not improve the selectivity. Simply lowering the reaction temperature to 23 °C (room temperature) led to an agreeable increase in enantioselectivity. However, further decreasing the temperature to 0 °C prolonged the reaction time without significant enhancement of the selectivity.

With the optimized conditions, we explored the scope of this variant of the aza-Claisen reaction. A number of stable, unprotected enamines containing cyano, ester, or nitro groups cleanly afforded the corresponding dihydropyridinones in good yields and enantioselectivities (Scheme 2). Similarly, a broad range of α,β -unsaturated aldehydes (aliphatic, alkenyl, or aromatic) proved to be suitable reaction partners. The best result (96% ee, entry 12) was obtained from coupling (*E*)-4-methylpent-2-enal and (*Z*)-3-aminobut-2-enenitrile. Less sterically hindered enals led to slightly diminished selectivity (entries 9 to 11). Aromatic groups either on the enals (entries 6 to 8) or the enamines (entries 16 to 17) afforded the expected dihydropyridinones in greater yield than the aliphatic counterparts. It is worth noting that the geometry of the

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enamine (either E or Z such as for entries 13 and 14) did not play a role in either reactivity or selectivity. In a preliminary attempt, α -methyl cinnamaldehyde did not participate in this reaction.

Scheme 2. Enantioselective Annulation of Enals and Enamines^{a,b}

^a Reaction conditions: 0.1 M PhCH₃, 12 h, rt. ^b(Z)-Enamines were used, unless noted. 'Yield refers to isolated yield after chromatography. ^dThe absolute configuration of (*R*)-**8** was established by X-ray analysis; others were assigned by analogy. ^eAn (*E*)-enamine was used.

We have previously acknowledged the relative difficulty of preparing cinnamaldehyde derivatives and introduced α' -hydroxyenones¹⁵ as easily prepared (one step from commercial materials) and stored surrogates. The increased sterics of these substrates preclude the use of chiral NHCs but smaller, achiral variants effect most of the known NHC-catalyzed annulations to give racemic products. In many cases, the synthesis of the racemic heterocycles is desirable for testing or reaction screening, and the enantioselective variant can be accomplished by preparing, when necessary, the corresponding enals. In order to test this new NHC-catalyzed annulation with a greater variety of aromatic and heteroaromatic substituents we developed

conditions for annulations of vinylogous amides and α' -hydroxyenones with achiral *N*-mesityl triazolium salt **18**. Consistent with our previous study, ¹⁶ triazolium catalysts lacking an ortho-substitution were found to be ineffective for this reaction (see Supporting Information for screening). The reaction proceeded cleanly and efficiently with these substrates and gave uniformly high yields from α' -hydroxyenones bearing electron-rich, electron-deficient, and hetero aromatic groups (Scheme 3). Again, a range of stable, unprotected enamines containing cyano,

Scheme 3. Annulation of α' -Hydroxyenone and Enamines^{a,b}

^a Reaction conditions: 0.1 M PhCH₃, 12 h at 40 °C, $^b(Z)$ -Enamines were used. b Yield refers to isolated yield after chromatography.

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ester, ketone (acyclic and cyclic), or nitro groups all participated in this reaction.

Two examples from Schemes 2 and 3 are worth noting. Compound **15** (Scheme 2), prepared in quantitative yield and high enantiomeric excess from the corresponding enal and a vinylogous amide, constitutes an asymmetric synthesis of a precursor for an α_{1a} adrenergic receptor antagonist with low nano- to picomolar potency, in which no other method for catalytic enantioselective synthesis is available. ^{4a} Similarly, compound **22** (Scheme 3) represents a facile catalytic synthesis of a precursor for selective Rho-kinase inhibitors ^{4b} from a readily prepared α' -hydroxyenone. ¹⁵

Scheme 4. N-Acylation of Enamines and Their Rearrangements

Mechanistically, we propose that the N-heterocyclic carbene, generated by deprotonation of the corresponding triazolium salt, adds irreversibly 16 to the enal to generate the Breslow intermediate I. This species is oxidized by 2 to form the key α,β -unsaturated acyl azolium II. Hickmott¹⁷ has demonstrated that the reaction of enamines such as 34 with α,β -unsaturated acid chlorides 35 proceeded first via N-acylation over C-alkylation, ¹⁸ which gave an intermediate that underwent a sigmatropic rearrangement, possibly via the intermediacy of a ketene (Scheme 4). Building on these results and those of Stork¹⁹ and Stille,²⁰ Mahalanabis²¹ has invoked this mechanistic rationale in the synthesis of 33 from acyl chloride 38 and enamine 39 (Scheme 4). In a slight departure from these precedents, we favor a reversible 1,2 addition of the enamine to II to form hemiaminal III, an argument that parallels our previous studies^{6,11} and the welldocumented properties of acyl azoliums. 12a-c

The metastable hemiaminal III is poised for an aza-Claisen rearrangement, the enantiodetermining step, in which the sense of asymmetric induction is identical to that observed in our dihydropyranone synthesis. This rearrangement is followed by lactam formation and catalyst turnover to afford dihydropyridinone products (Figure 1).

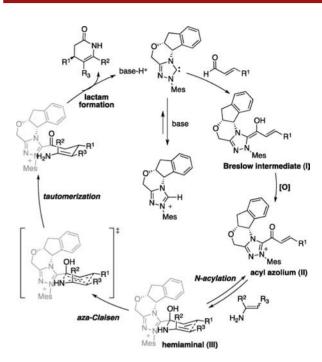


Figure 1. Proposed mechanism of an NHC-catalyzed aza-Claisen annulation cascade.

In summary, we have reported the annulation of enals or their α' -hydroxyenones surrogates with readily prepared vinylogous amides via an NHC-catalyzed aza-Claisen rearrangement. The reaction proceeds with a broad range of substrates to afford synthetically useful dihydropyridinones in good yields and enantioselectivities. This reaction also constitutes an NHC-catalyzed N-heterocycle formation, in which neither imines nor nitrogen protecting groups are required.

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Supporting Information Available. Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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